# Activities of Lipoprotein Lipase and Hepatic Lipase on Long- and Medium-Chain Triglyceride Emulsions Used in Parenteral Nutrition

Olivier Lutz, Thierry Lave, Anny Frey, Zahia Meraihi, and André C. Bach

Protonged parenterel nutrition frequently includes lipid emulsions. This report investigates how emulsions containing traingly-greated of different molecular weight affect the rate of clearance in vivo and the activity in vivo of the two enzymes responsible for this clearance: disphragm lipoprotein lipses (LPL) and hapatic endothelial lipses (LRL). Whatever their molecular weight, the trizerigylacerols of the emulsions were hydroyaded by LPL and All. However, the reaction was fear with medium-chain trighycerides (MCT) than with long-chain trighycerides (LCT). To be active, LPL required the presence of summing the remaining t

IN MAMMALS, circulating triacylglycerols cannot cross biologic membranes. For these neutral lipids to move from their carrier (chylomicrons, very-low-density lipoproteins [VLDL], or fat droplets that have been inflused in the form of a phospholipid-stabilized emulsion) into the tissues, they have to be hydrolyzed with the release of fatty acids, they have to be hydrolyzed with the release of fatty acids. This hydrolysis is performed by two enzymes: lipoprotein lipases (LPL; EC 3.1.1.34), which is located at the capillary modothelium in extrahepatic itsues, and bepatic lipase (HL; EC 3.1.1.33). The hydrolytic activity of these two enzymes provides the major pathway by which endogenous or exogenous triacylglycerol-rich particles are cleared from the bloodstream.

Parenteral nutrition frequently includes a supply of lipids in the form of an emulsion. At present, two types of emulsion are commercially available, distinguished by the nature of the oil used. It is generally a vegetable oil (soybean or safflower oil; LCT) the constituent triacylglycerols of which have long-chain fatty acids (longer than C12, LCFA). Recently, emulsions have become available that provide medium-chain triglycerides (MCT) (constituent fatty acids between C6 and C12, MCFA). The specific physicochemical properties of MCT and MCFA3 have led more and more clinicians to introduce this type of emulsion into the parenteral nutrition of their patients.4 Our aim here was to investigate in rats how the nature of the oil used to make the emulsion affects the rate of clearance of this emulsion in vivo and the activity of the two enzymes responsible for most of the clearance.

### MATERIALS AND METHODS

Intravenous Fat Tolerance Test

Between 1 and 3 PM male rats (OFA strain, from lifts-Credo, Ivabreale, France), weighing approximately 400 g, were anesthetized with sodium pentobarbital (40 to 50 mg/kg body weight; Clin Midy, Saint lean de la Rullel, France). A catheter (Intramodie PE 50, RUA Instruments, Torry, France) was introduced into the jugular voin and another (Silisatic medical grade tubing, Dow

artey. The first catheter was maintained with a 9 g/L NaCl initiason (4.5 m/L, for 15 mituse before and 1.2 m/L, hat the fat infusion) and the second filled with a citrate buffer, pH 7.4. The animals were not given heparin. The fipid emulsion was infused through the jugular vein (12 m/L/h) (B. Braun Perfusor. Melsungen. West Germany) for four minutes. The smount of lipids represented 0.8 mL of emulsion. For rats weighing approximately 400 g, this represents approximately 0.4 g tries (14 m/L/h) (3.6 m/L) (4.6 m/

Immediately after the blood sample was collected, part of it was uslably diluted\* in mannitol (50.7 g/L) mixed carefully, and centralged (620  $\times$  g for 15 minutas) (2RD centrifuge, Sigma, Osterode am Harz, West Germany). The supernatum was transferred to a cuvette, and its light-scattering index (LSI) was determined (arbitrary units) at 650 nm as soon as possible by means of a nephelometer (Photovolic Corporation, NY).

This method of measuring the light scattered by the suspended fat particles made it possible to evaluate the amount of exogenous lipids present in the circulating blood. From the measurements taken each five minutes, we deduce the value at 0 minutes. The net values follow a straight lime when plotted on a semilograftimic scale.

The rest of the blood was also diluted and then centrifuged (a 600 v g for 15 minutes), and the plasma was decented and froze on at  $-20^{\circ}$ C until the moment of estimation of free glycerol' and triacylglycerol. (Triglyceridae norymatique. Triader, Biotriol, Paris, France, free glycerol is substrated from total glycerol). Monesterifield flaty acid (NEFA) levels were measured by an enzymatic method (Wako Nefa C test, Biolyon, Dardilly, France, which is applicable to MCFA. The expression of the calibration curve of  $C_{\rm sc}$  is  $y=0.95\times-1.92$ . Hematoori determinations permitted the

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expression of the results obtained in terms of concentration per liter of plasma. At times 0 and 40 minutes, 600  $\mu$ L of whole blood was precipitated with HClO<sub>4</sub> for estimation of the  $\beta$ -hydroxybutyrate.<sup>9</sup>

### Measurement of the Activity of Diaphragm Lipoprotein Lipase

The activity of the LPL from rat disphragms was measured using an adaptation of Boulang's method. "The disphragm was removed from freshly desapitated male rats, quickly rinsed in NH<sub>2</sub>Cl, dried, and then ground for 15 seconds in an Ultra-Turrar gridner, Janke and Kunkel, Stauten, West Germany) in cold NH<sub>2</sub>Cl/peparin (Sconnising 4 Ul heparin/ml. 1 g tissue per 12 ml. NH<sub>2</sub>Cl/peparin). After 15 minutes of contact at 4°C, the lower than the state of the sta

To measure the LPL activity, 0.3 vol of a 20% emulsion was presented for 30 minutes at 4°C with 1.2 vol of inactivates at 4°C with 1.2 vol of allowing 1050 xff, pH 8.6; fraction V, Sigma, St Louis) and 0.9 vol of glycine (100 mmo/L<sub>L</sub>, pH 8.6) were added to the incubation medium. After addition of 5.0 vol of disphragm extract, the incubation medium was stirred and heated to 37°C.

At various times, the reaction was stopped with ice-cold bettemenporonic acid (final concentration O.8 mmo/J.). (EgA.-Chemic, Steinheim, Aldrich, West Germany). In The misture was centrifuged a vivice (two misuses at 9,000 × gr., 2000 centrifuge, Eppendorf, Hamburg, West Germany) to climinate lipid particules by forsation. On the underlying liquid we estimated in deplicate the free farty acids. Benzeneboronic acid had no effect on the estimation of fasty scide per se.

The release of free fatty acids was linear with time for at least 60 minutes. A control incubation, in which the tissue extrict was replaced with NH<sub>2</sub>C-heparin was run in parallel with each experient. The measured activity, expressed as small free fatty acids released/g, muscle/h, was taken to be LPL activity because it was inhibited by either the omission of serum Apo G<sub>n</sub> the addition of NaCl, or of protamine sulfate (Serva, Heidelberg, West Germany) as shown by others. <sup>154</sup>

#### Measurement of Henatic Lipase Activity

Male rats were anesthetized and their livers were perfused (Waten-Mardow perfuser, Falmouth, Cormsul, England) in situ according to the method of Henns et al. \*\* The liver was rinsed initially with 200 ml. Krebs-Henselsti bicarrbonate buffer, plf 7.4, over a period of about ten minutes, without recycling. It was then perfused for six minutes with Krebs-Henselsti/Reparin (1 m.L/10) at 60 ml. buffer/gs body weight. This time the perfusate was recycled, and its was then centrifuged and divided into aliquots that was the method of the perfusate was recycled, and its was then centrifuged and divided into aliquots that

added 2.6 vol of bovine albumin (190.5 g/L, pH 8.6), 6.3 to 6.7 vol of 100 mmol/L, gb/mc (pH 8.6), and 0.4 to 0.3 vol enzyme suspension. The procedure for tening HL activity was the same as that for LPL. The release of free first yacids was linear with time for 6.0 minutes. In a parallel control test, the enzyme extract was heated in advance to a parallel control test, the enzyme extract was heated in advance to 100°C for two minutes. The HL cativity was expressed as micromodes of free fatty acids released/g liver/h. NaCl and protamine sulfate had tittle effect on HL activity. 30.151

### Emulsions

Table 1 shows the compositions of the various emulsions used.

Structured lipids (SL-MCT/LCT) are randomly esterified triacylglycerols of MCFA and LCFA.

#### RESULTS

#### Introvenous Fat Tolerance Test

In our experimental conditions we obtained a biphasic elimination curve (Fig 1), indicating zero-order kinetics for removal at high fat concentrations and first-order kinetics at lower substrate concentrations. With the change from LCT to MCT, the clearance rate increased and the rise of Bhydroxybutyrate (between 0 and 40 minutes the increase was, respectively, +209 ± 21 μmol/L for LCT v +702 ± 50 for MCT) was greater. The increase in circulating lipids was also greater (Fig i). If with LCT, and to a lesser extent with MCT, the curve of LSI variation appeared as being more abrupt than those of plasma lipids, it is explained both by the difference in the expression of results (as concentration in the case of LSI; as percentage of infused fat in the case of plasma lipids) and by the intervention in the clearance of a supplementary mechanism, the uptake by the reticuloendothelial system.

A comparison of the results obtained after the infusion of a 20% 70/30 (wt/wt) MCT/LCT emulsion with those obtained after infusion of a 70/30 structured lipid emulsion (Fig 2) shows that the clearance rates are identical as are the changes in plasma lipids and blood  $\beta$ -hydroxybutyrate ( $4+25\pm40$   $\mu$ mol/L for MCT/LCT  $\nu$   $+408\pm29$  for SL-MCT/LCT) levels.

#### In Vitro Experiments

An MCT emulsion was more rapidly hydrolyzed by LPL or HL than was an LCT emulsion (Fig 3). In the case of a mixed MCT/LCT emulsion, the activity of the two enzymes went up as the percentage of MCT increased (Fig 4). For LPL the V<sub>max</sub> and K<sub>m</sub> were, repectively, 70 µmol/g/h and 3.67 mmol/L for the 20% LCT emulsion, and 154 µmol/g/h

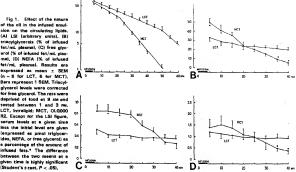
Table 1. Composition of the Emulsions Used

Lipids	Trade Name or Batch Number	Menufecturer	Oils (g/L)
LCT	Intralipid 20% or 10%	Kabi-Vitrum	Soybean oil (200 or 100)
	Endolipide 20% or 10%	Bruneau	Soybean oil (200 or 100)
MCT/LCT	Lipofundin MCT, 50/50 20% or 10%	B. Braun Melsungen	Soybean oil (100 or 50) + MCT (100 or 50)
	MCT/LCT, 70/30, 20% (V. 12 274)	B. Braun Melsungen	Soybean oil (60) + MCT (140)
	SL-MCT/LCT 70/30, 20% (V. 12 275)	B. Braun Melsungen	Soybean oil (60) + MCT (140); Structured lipid:
MCT	MCT 20% (CH E 86024)	B. Braun Melsungen	MCT (200)
	MCT 20% (OLGOOD R2-PXE0815)	Travenol	MCT (200) -

Egg phosphatides (12 g/L) ware always added as an emulsifier and olycerol (22.5 g/L for Intraficid: otherwise 25 g/L) as an isotonic adjuvant.



Student's t-test, P < .05).



and 3.55 mmol/L for the 20% MCT emulsion. For HL these values were, respectively, 15.5 µmol/g/h and 3.27 mmol/L for the 20% LCT emulsion and 18.2 µmol/g/h and 0.61 mmol/L for the 20% MCT emulsion. The activity of LPL, similar to that of HL, increased with increasing amounts of albumin in the incubation medium; maximum activity was reached more quickly with MCT than with LCT (Fig 5A,B). The same was true with the inactivated serum in the case of LPL (Fig 5C,D). The activity of HL toward LCT decreased greatly in the presence of serum, whereas toward MCT it increased slightly. The hydrolysis of structured lipids was slightly slower than that of the corresponding MCT/LCT

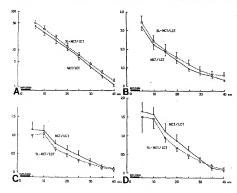


Fig 2. Effect of a physical v e chemical (structured) MCT/ LCT mixture on the circulating lipids. (A) LSI (erbitrery units). (B) triecylglycerots (% of infused fat/mL plesme), (C) free glycerol (% of infused fat/mL plesme), (D) NEFA, (% of infused fet/mL plesme). Results are expressed as mean ± SEM (n - 9 for MCT/LCT, and 8 for SL-MCT/LCT). Bars represent 1 SEM. Triecylglycerol els were corrected for unesified glycerol. The rets were fasted, MCT/LCT, V.12 274; SL-MCT/LCT, V.12 275. Except for the LSI figure, serum levels et a given time ss the initial level ere given (expressed es µmol triglycerides, NEFA, or free glycerol) as a percentage of the amount of

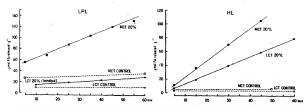


Fig 3. Hydrolysis of fat emulsion by LPL and HL as a function of time. FA, fatty ecids. LCT 20 %, Endolipide; MCT, OLGO00 R2. Incubation mixtures: LPL: 30 μL 20% emulsion; 260 μL albumin; 120 μL as a run; 30 μL dylorie; 500 μL enzyme in NH<sub>4</sub>Cl/haparin. HL: 30 μL 20% emulsion; 260 μL albumin; 660 μL glycine; 50 μL enzyme in Krébs-Henseleit buffer.

mixture (Fig 6). With LPL the K<sub>m</sub> was 8.51 mmol/L with the structured MCT/LCT (SL-MCT/LCT) preparation and 4.78 mmol/L with MCT/LCT, whereas the V<sub>mm</sub> values were, respectively, 98 and 117 µmol/g/h. For HL the respective values were K<sub>m</sub> 1.23 and 1.57 mmol/L and V<sub>ms</sub>. 12.6 and 15.4 µmol/g/h.

#### DISCUSSION

### Long- and Medium-Chain Triglycerides

It is well known that LPL hydrolyzes triacylglycerol-rich ipportesias? and particles of artificial fat emulsions. Shall insportesias? and particles of artificial fat emulsions shall clikewise, HL undeniably shows a triacylglycerol lipase activity in vitro. 3The results shown in Fig 3 point in the same direction: the LCTr. or MCTr-based fat emulsions, are hydrolyzed in vitro, both by LPT2 and by HL. These results also widen the applicability of the observation made by Wang et al? he human milk LPT2 muscle PLP hydrolyzed MCT emulsions more rapidly than it did LCT emulsions. The same was true of HL (Fig. 3). Mixed MCT/LCT

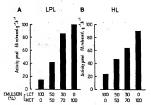


Fig. 4. Effects of the relative quantities of MCT and LCT in the 20% enuties on the activity of EPL (a) and of HL (B). Fa. firstly ecids; LCT, Endelipide; MCT/LCT 80/50, Lipofundin MCT; MCT/ LCT 70/30, V.12 274; MCT of E 88 024; Rucubestion mixtures: LPL: 30 µL 20% enuties; 120 µL serum; 260 µL albumin. 30 µL glycine. 500 µL anzyme in NH\_CI/ heparin. HL: 30 µL 20% emulsion; 260 µL abbumin: 70 µL enzyme in Krebs-Henselst biffer; 69 µL glycine.

emulsions lay between LCT and MCT (Fig 4). The release of fatty acids by LPL and HL became faster as the percentage of MCT in the mixture increased.

The faster hydrolysis of MCT by LPL is confirmed by the Vms value, which was 2.2 times larger for MCT than for LCT. The ratio Vmax MCT/Vmax LCT was only 1.2 in the case of HL. With regard to the LPL affinity (1/K\_), it is the same for both substrates (ratio MCT/LCT = 1.0); HL affinity, on the contrary, was widely in favor of MCT (ratio 5.4). Several hypotheses can be proposed to explain these results: (1) The ester bond is weaker in MCT.1 (2) The interfacial tension of oil against water being less in the case of MCT than for the LCT, the MCT based emulsions are more fragile. The faster destruction of the MCT in the presence of NaCl is consistent with this fact. (3) A difference in the surface area between the two types of fat particles allows the enzyme to better hydrolyze the MCT-based emulsions. (4) The existence of a retroinhibition of enzymes by the released fatty acids would be stronger with LCFA.

### Structured Lipids

High hopes are held for structured lipids,25 emulsions of which are thought to bring about a positive nitrogen balance in parenterally fed rats more rapidly than the currently used commercial emulsions.26 In a 70/30 (wt/wt) MCT/LCT emulsion each lipid particle contains molecules of both MCT and LCT. Because their mean molecular weights are, respectively, 492 and 872, 81% of the triacylglycerol molecules possess an MCFA in position sn-1 (as well as in positions 3 and 2). This is not the case with structured lipids. These triacylglycerols contain in the majority both MCFA and LCFA in the same molecule. Considering the initial proportions (70/30 wt/wt) and the random distribution of fatty acids, it is safe to say that the majority of triacylglycerol molecules of the SL-MCT/LCT emulsion contain two MCFA for one LCFA. Because in certain of these triacylglycerols, the LCFA occupies the position sn-1, one can assume that approximately 66% (v 81% in MCT/LCT) of the MCFA are in a favored position for action by the LPL.1 We observed that these emulsions were hydrolyzed by the two lipases involved in the catabolism of triacylglycerol-

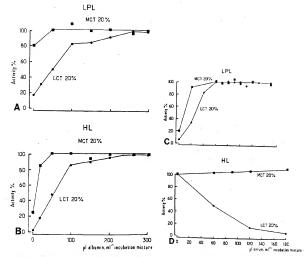


Fig. 5. Dependence of LPL and ML activity on the albumin (A, B) or inactivated serum (C, D) concentrations in the mixture. LCT, interligicity MCT, CE B 80 c4. The observed activity vitor 300 st. of albumin (A, B) and 720 pt (for LPL) — 0 pt (for ML) areum (C, D) was arbitrarily taken as 100%. Incubation mixtures: LPL: 30 pt 20% on the control 100 kJ or 0-100 (C) pt assure, 0-30 0l arrayme in McJ (C) began't in the medium was made up to 1 mL worth givine solution. 11. 30 pt 20% emission: 0 (B) or -100 (D) pt assure, 0-30 (B) or -200 (B) or -

rich particles, but slightly more slowly than the usual MCT/LCT emulsions (Fig. 6). The V<sub>max</sub> ratio for MCT/LCT v. St.-MCT/LCT was 1.2, ie, identical to that of \$1/65, the ratio of the MCFA molecular quantities in the favored position in the two emulsions. LPL had a higher affinity for the physical mixture (ratio MCT/LCT v SL-MCT/LCT 1.3), whereas HL seemed to prefer the structured ligids (ratio 0.8).

# Albumin Requirement

Because fatty acids released by the enzymatic action on an LCT emulsion are not water-soluble, serum albumin must be added to bind these components and to allow hydrolysis to proceed. This is indeed the case with LCT, both for LPL $^{1,1273}$  and HL $^{12}$  (Fig 5A,B). It is also true of MCT, but in this case less albumin is needed, and the maximum activity

is reached with lower concentrations of albumin. This observation may be explained by the difference in solubility of the various fatty acids in water: at 20°C the solubility is 7.2 mg/L (28 µmol/L) for palmitic acid \(^{\text{b}}\) 680 mg/L (4.716 µmol/L) for octanoic acid.\(^{\text{b}}\) Tributyrin is hydrolyzed by milk LPL in the absence of albumin.\(^{\text{b}}\)

## Serum Requirement

For maximum activity, L.P.L. requires the presence of apolipoprotein C<sub>II</sub>. Intablanta II an articla emulsion is used, it must first be activated by incubation with serum. Figure SC,D shows that the rate of triacylglycerol hydrolysis by L.P.L. increased as a function of serum concentration and that the maximum activity of L.P.L. on MCT is reached with less serum and hence less app.C<sub>II</sub> than is needed for L.CT. It had been shown previously that the enzymatic hydrolysis of

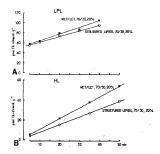


Fig 6. Hydrolysis of MCT/LCT structured lipids by LPL (A) and HL (B) FA, fatty acids; MCT/LCT, V. 12 274; structured lipids, V.12 275, incubation mixtures are as indicated in Fig 3.

tributyrin, is not apo $C_{\rm H}$ -dependent. <sup>233,20</sup> Although apo $C_{\rm H}$  is a specific activator for LPL, it is an inhibitor of HL. <sup>235</sup> We also observed that serum strongly inhibits HL, but only in the case of the LCT emulsion (Fig 5C,D), and not at all with the MCT emulsion. It is thus probable that in vivo the liver plays an important role in clearing MCT-based fat emulsions. The hyperketonemia - observed after MCT infusion - can be explained partially by this fact.

### Fate of Infused Fats

The difference in activity between LPL and HL toward MCT and LCT emulsions explains what is seen in vivo (Fig I): the former are cleared more rapidly than the latter, as has previously been shown in metabolically healthy adults.

and in dogs.38 Although MCT were cleared more rapidly than LCT, our results show that the plasma concentration of triacylglycerols increased more with MCT than with LCT. 36,39,40 This happens because for a given weight of emulsion, the two types of lipids do not provide the same number of fat molecules. Considering the mean molecular weight of the triacylglycerols used, the MCT provide approximately 1.8 times more triacylglycerol molecules than the LCT. The same explanation applies to glycerol, the plasma concentration of which (Fig. 1) increased the most following infusion with MCT. The increase in NEFA following the infusion of MCT did not differ significantly from that following LCT, but the difference was significant during the disappearance phase of the curve; at 35 and 40 minutes the NEFA were lower for MCT than for LCT. One explanation is that the half-life of MCFA is shorter than the half-life of LCFA.

During an MCT infusion, we theorize a larger role of liver in exogenous fat clearance, and observe a higher increase of circulating fatty acids. Since the work of Scheig, 41 it has been known that in the liver MCFA, unlike LCFA, are poorly incorporated into lipids and are strongly oxidized. The resulting increase in acetyl-CoA is greater with MCT than with LCT.42 This substantial β-oxidation of MCFA may explain the great increase in blood ketone levels after infusion of an MCT-based emulsion.36,37,39,40 Our results show that the amount of circulating  $\beta$ -hydroxybutyrate rose more with MCT than with LCT. In vivo the 70/30 MCT/LCT and 70/30 SL-MCT/LCT emulsions behaved indistinguishably. Clearance and changes in plasma lipid and blood β-hydroxybutyrate concentrations26 were the same (Fig 2), whereas in vitro the activity of the two enzymes involved in their clearance was a little greater with the "physical" mixture than with the "chemical" mixture (Fig 6). Because LPL and HL hydrolyze an MCT-based emulsion rapidly, these triacylglycerols are cleared more rapidly than an LCT emulsion. MCFA, which, not being stored, are energy substrates par excellence, are therefore more rapidly made available to the tissues of a living body than are the LCFA of a traditional emulsion.

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